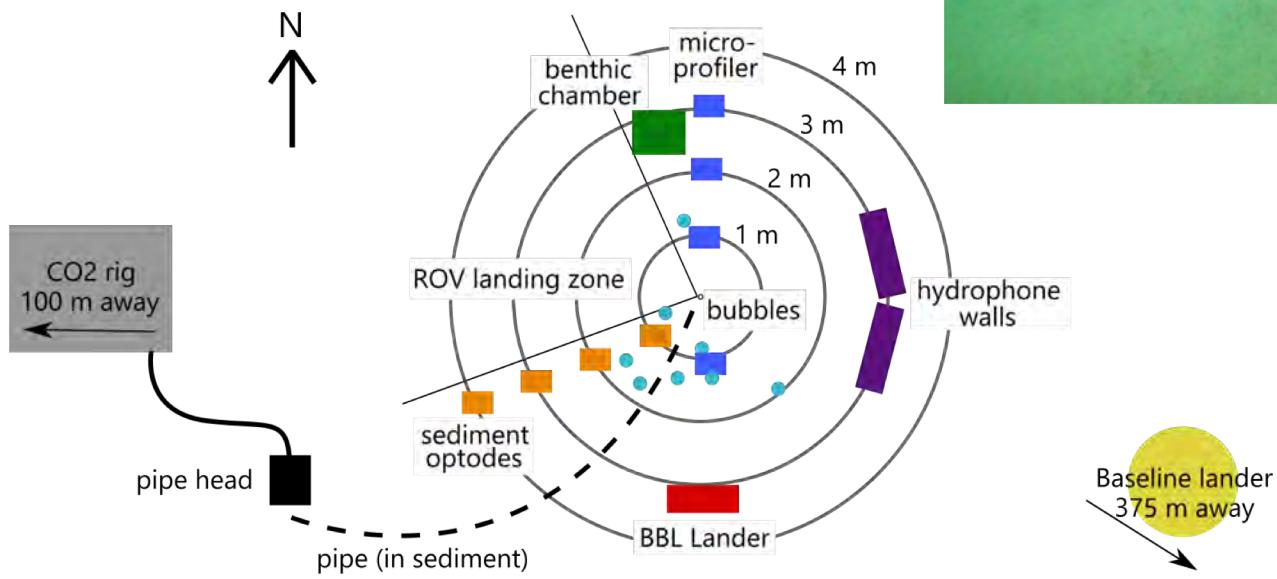
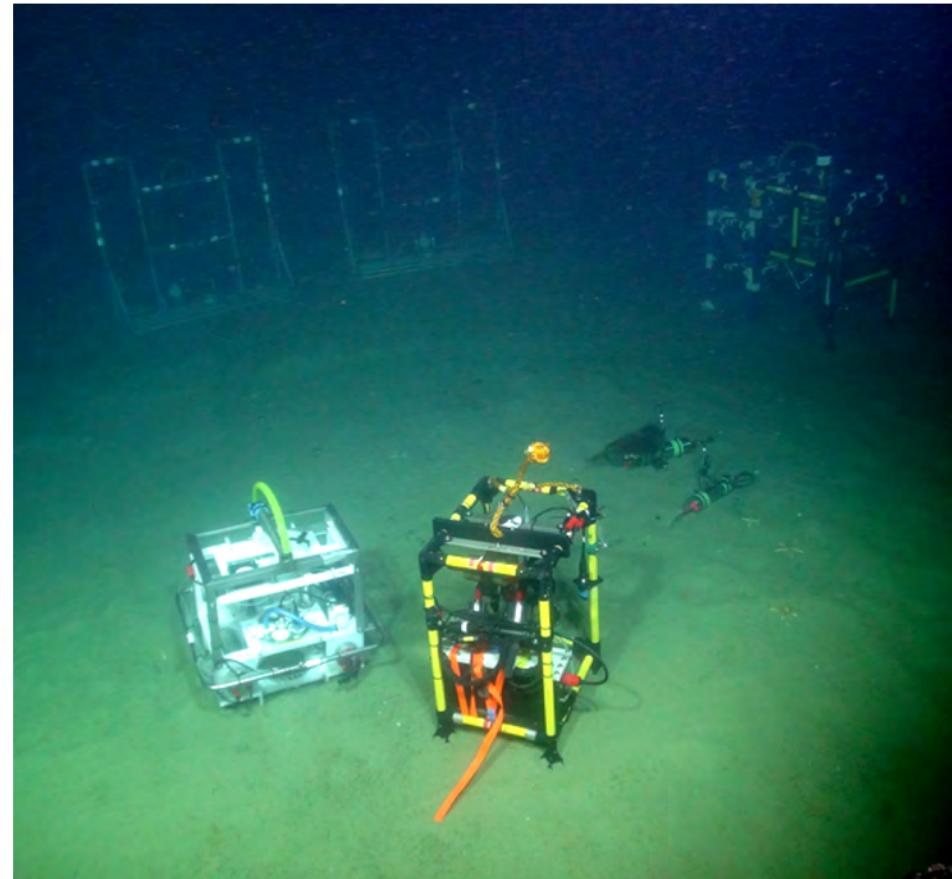
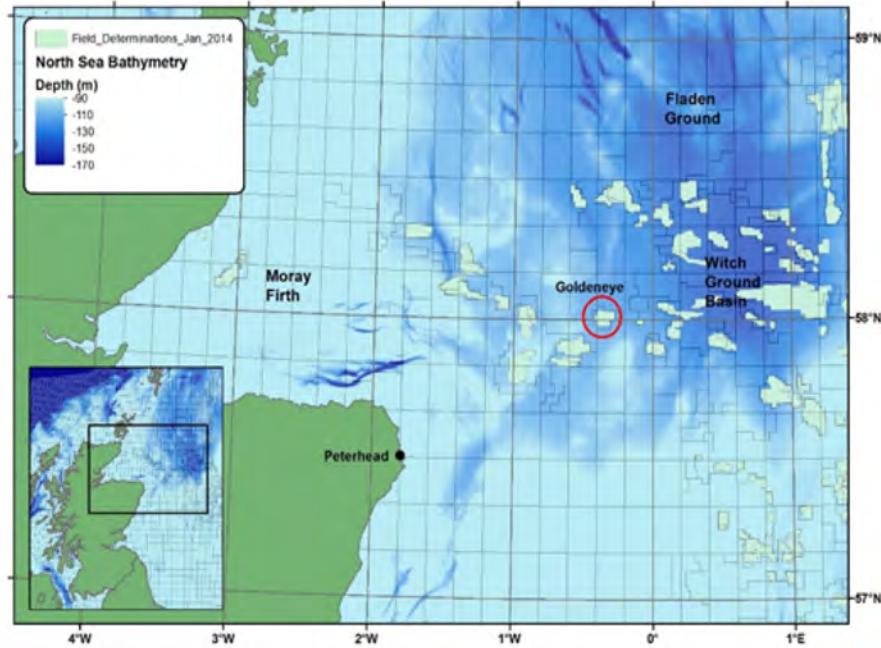




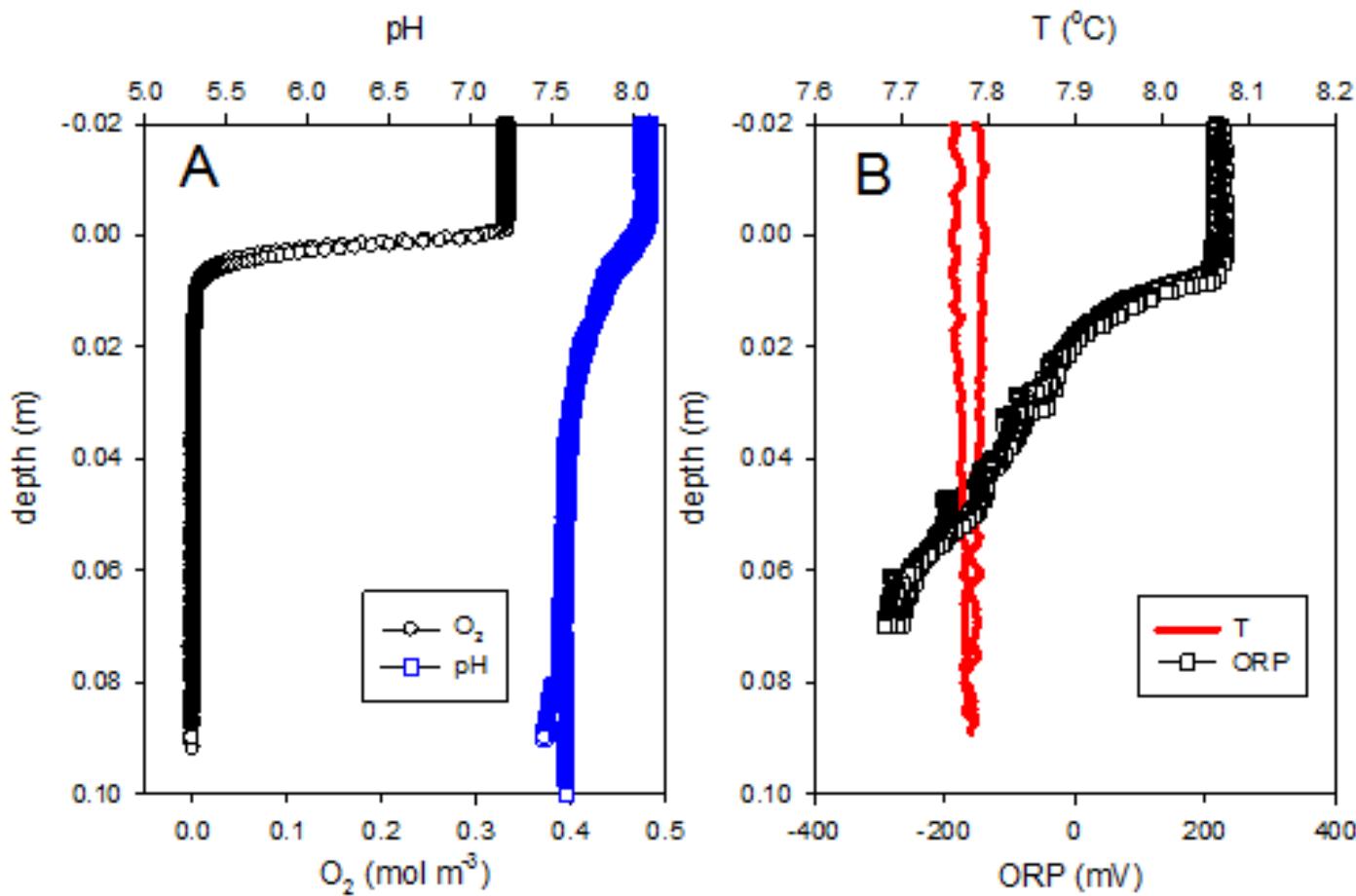
# Structure of vents using high resolution studies

Dirk de Beer, Anna Lichtschlag, Anita Flohr, Kate Peel,  
Dirk Koopmans, Moritz Holtappel

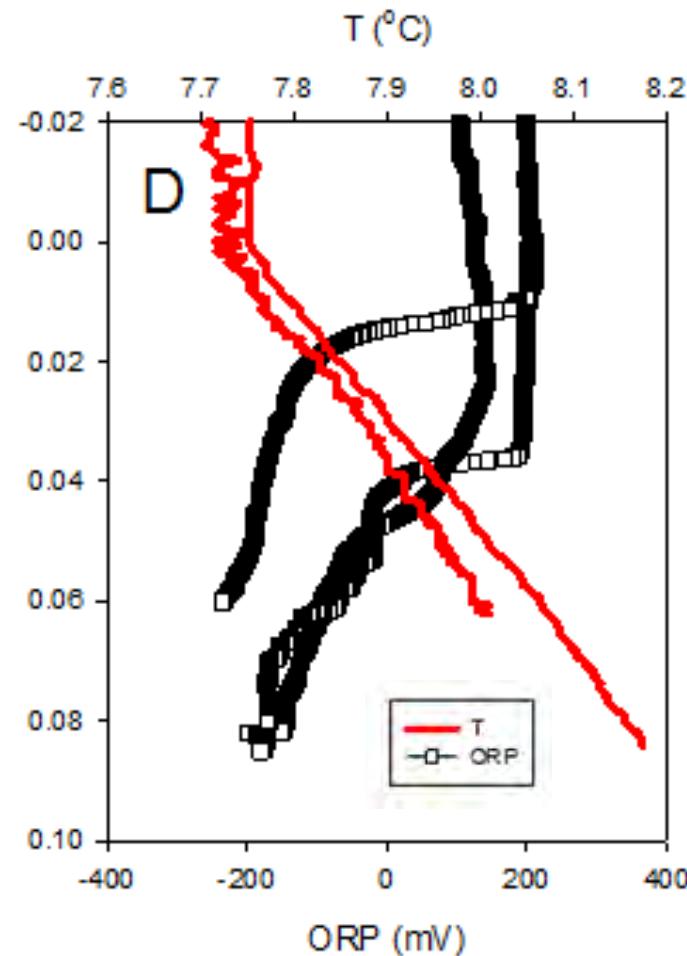
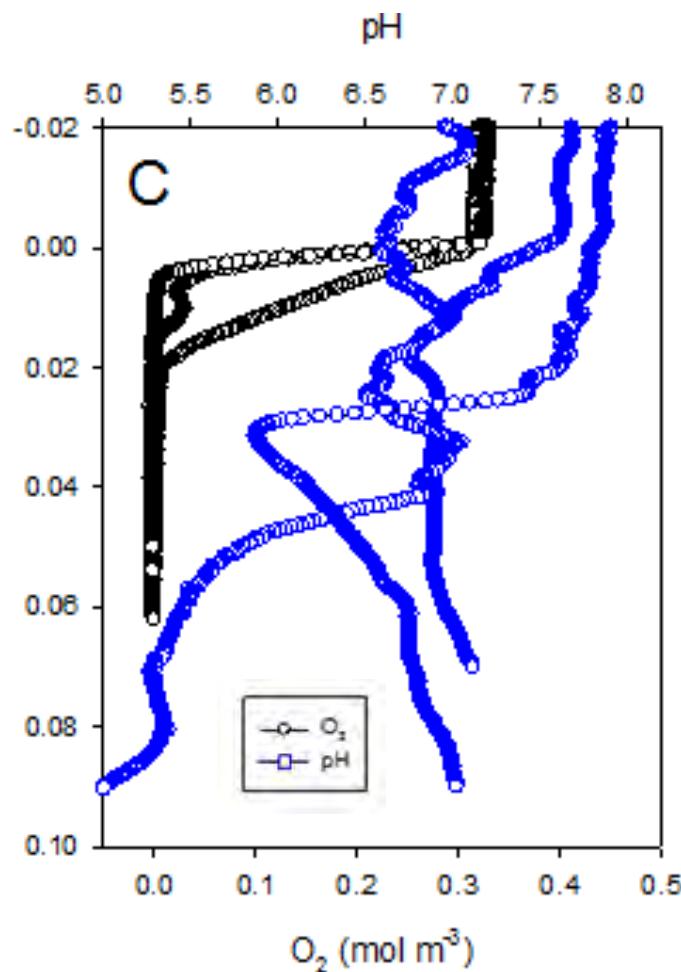
MPI-NOCS-AWI

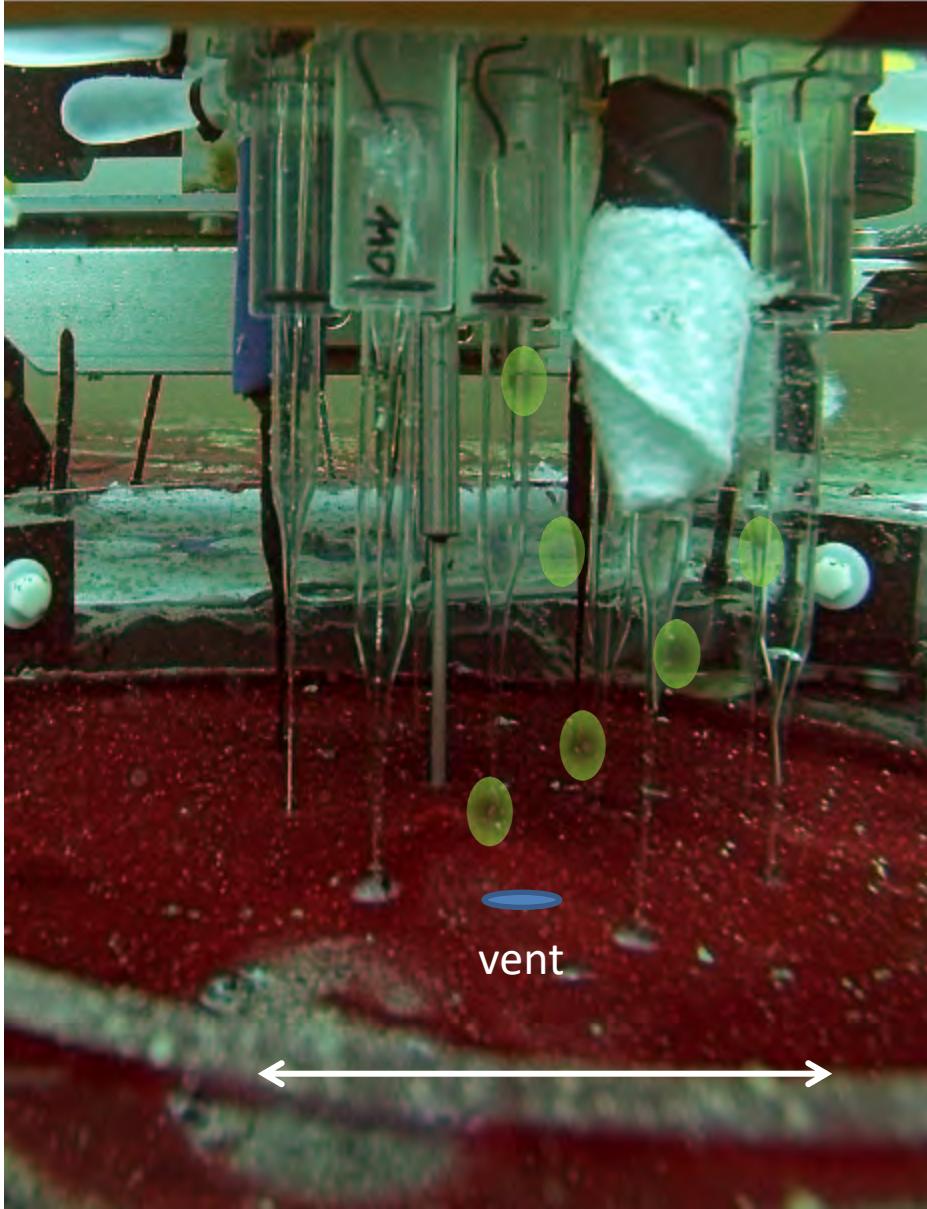


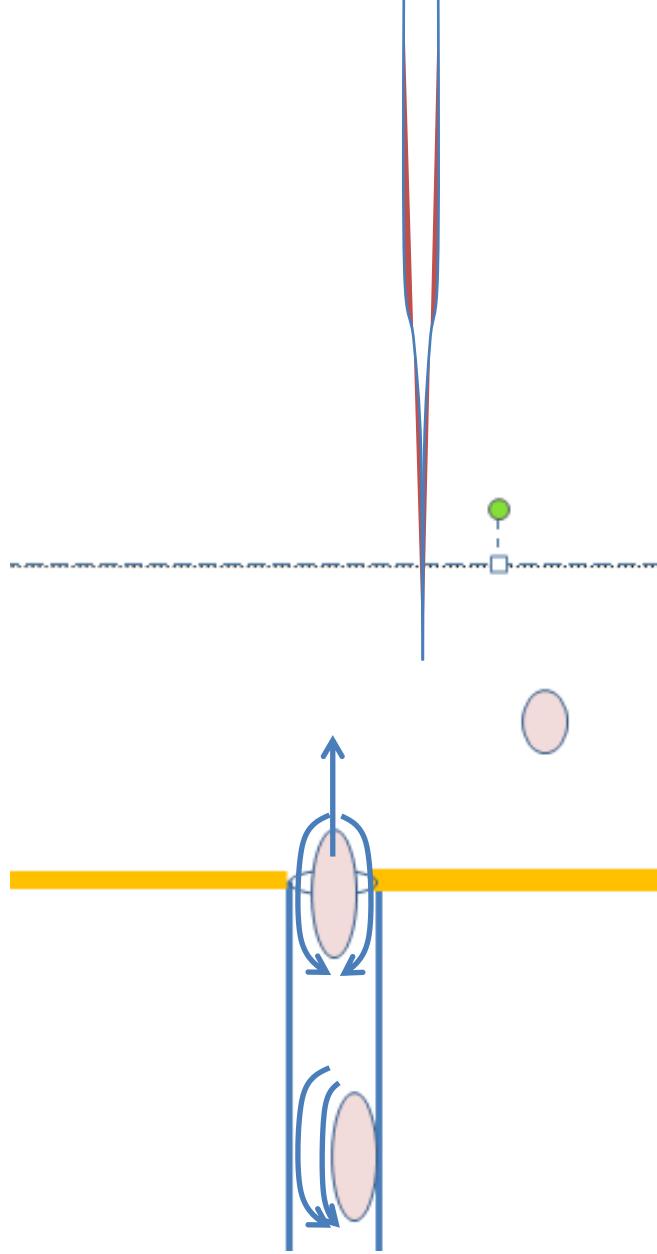
# Background profiles



# Profiles in vent





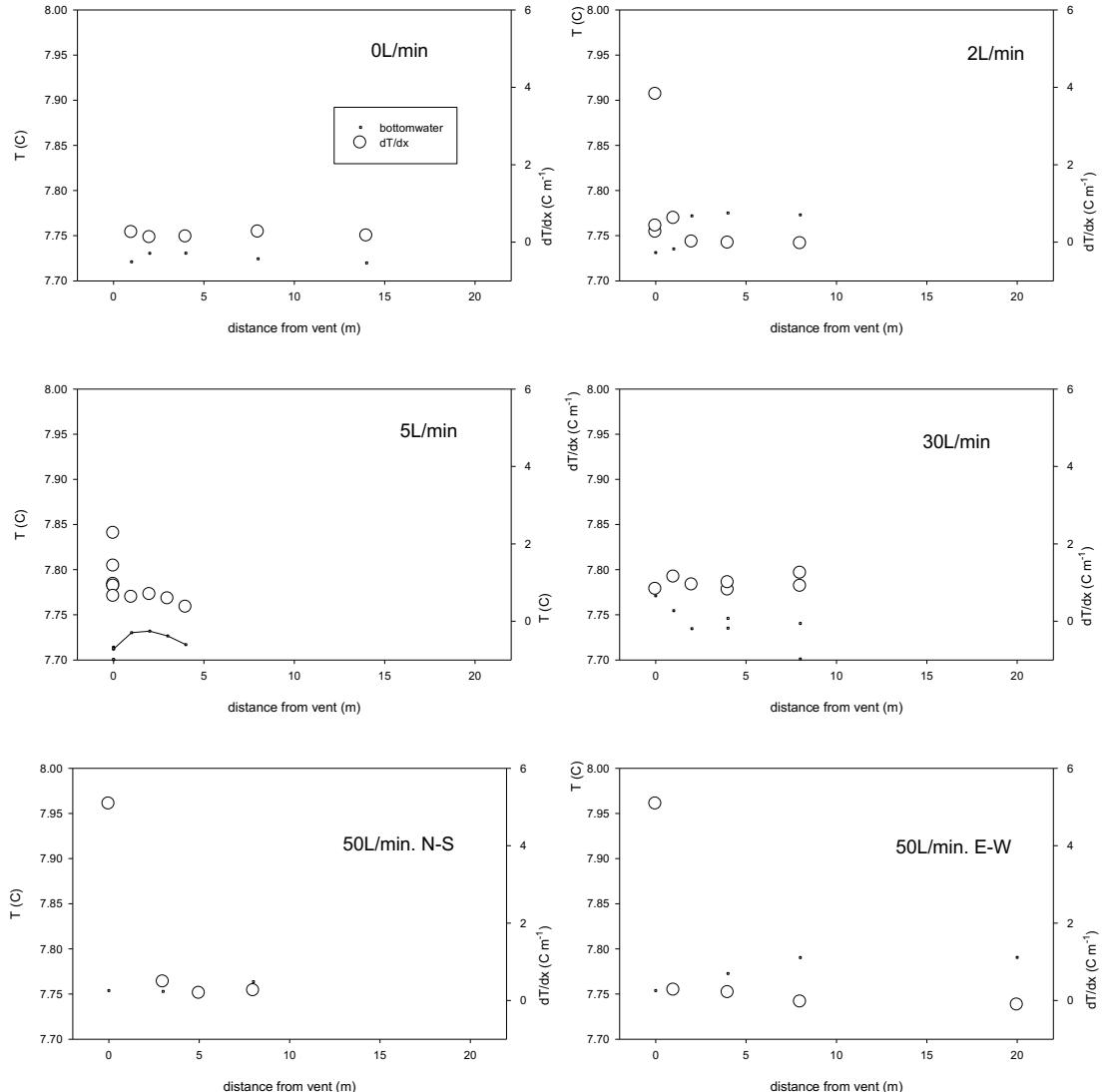


# T transects

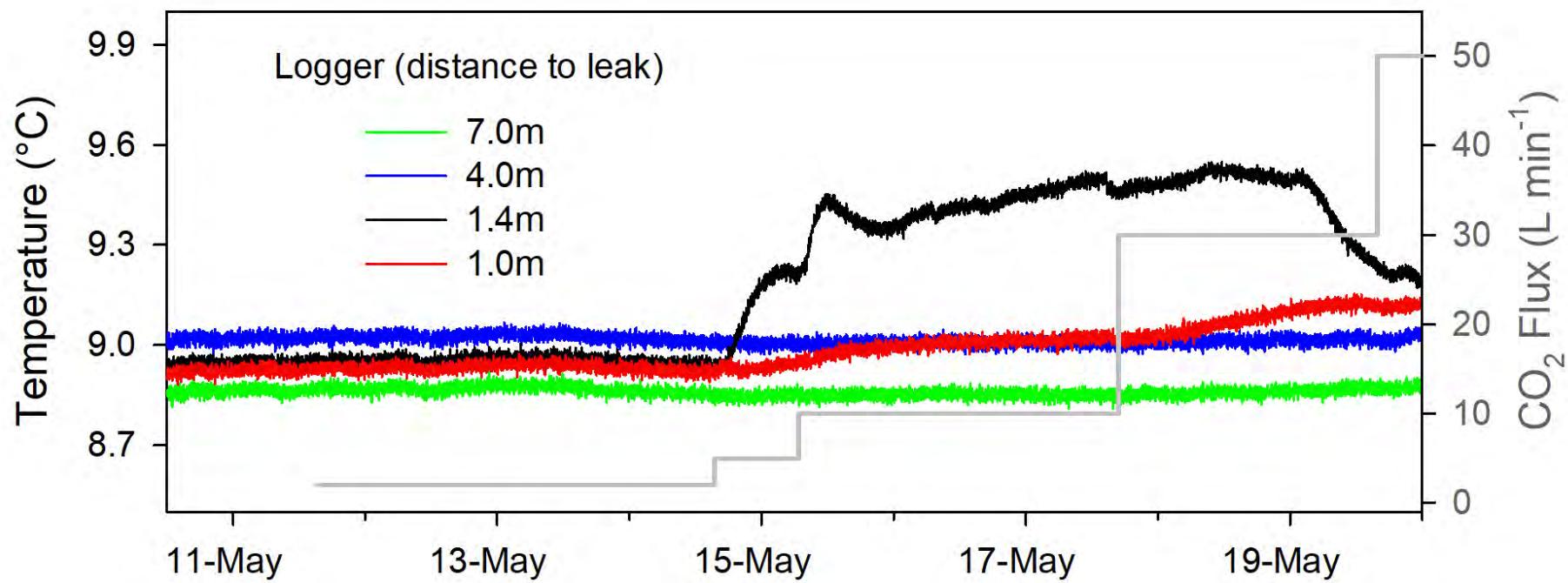
T-gradients local  
(<1 m from vent)

Not proportional  
to CO<sub>2</sub> release  
and variable

No T in bottom  
water



# Heat generation



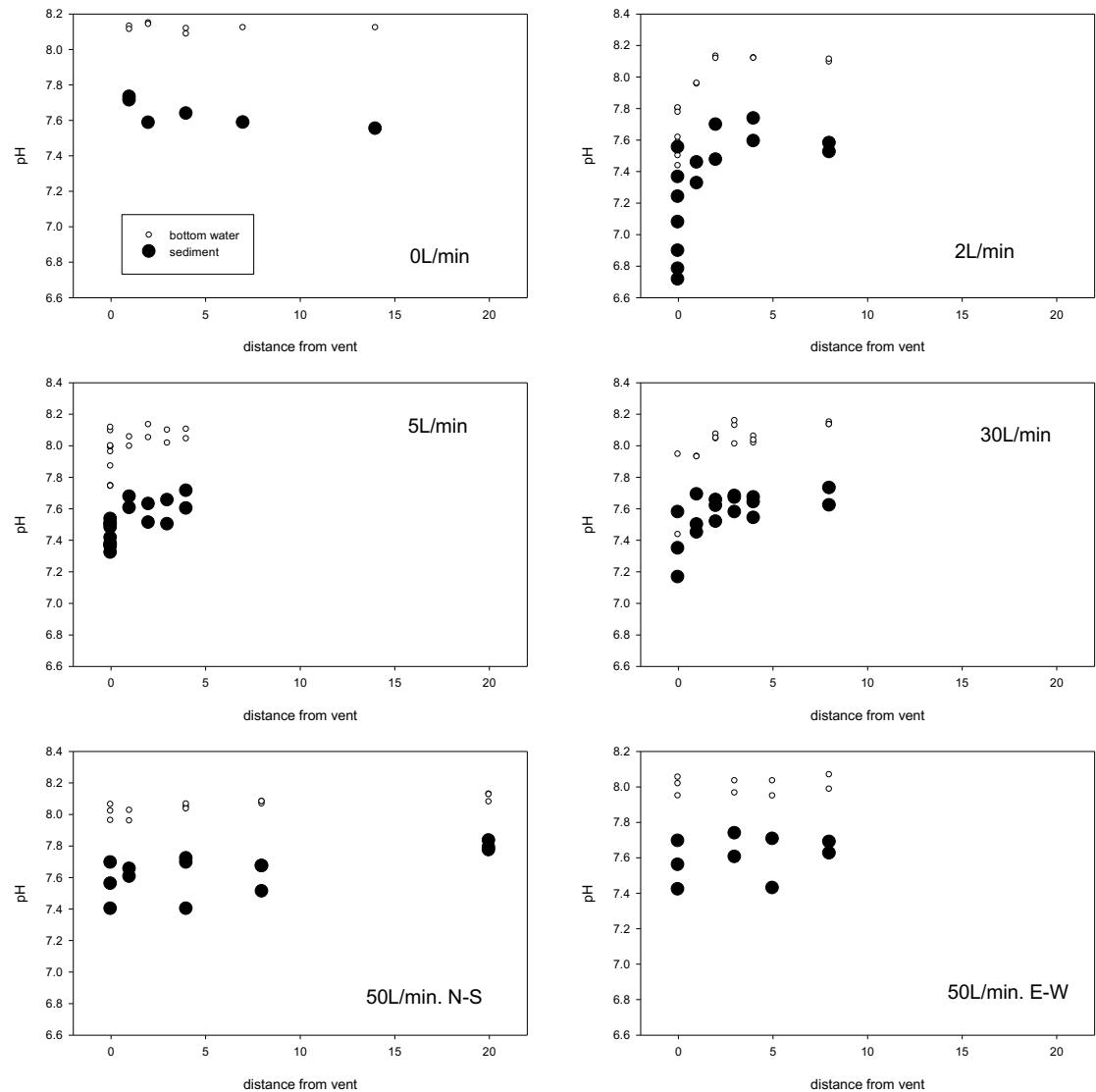
Loggers from Moritz Holtappel, 30 cm depth

# pH

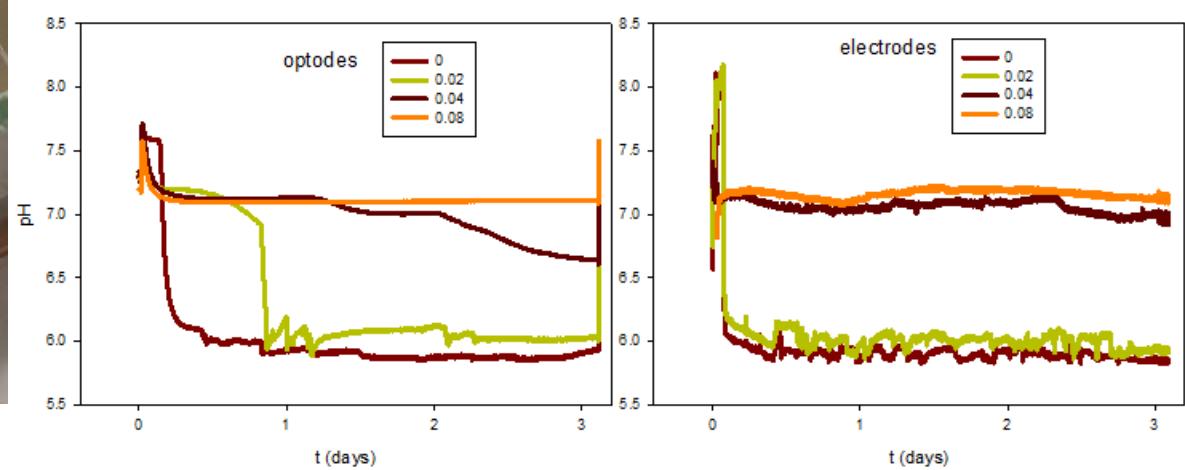
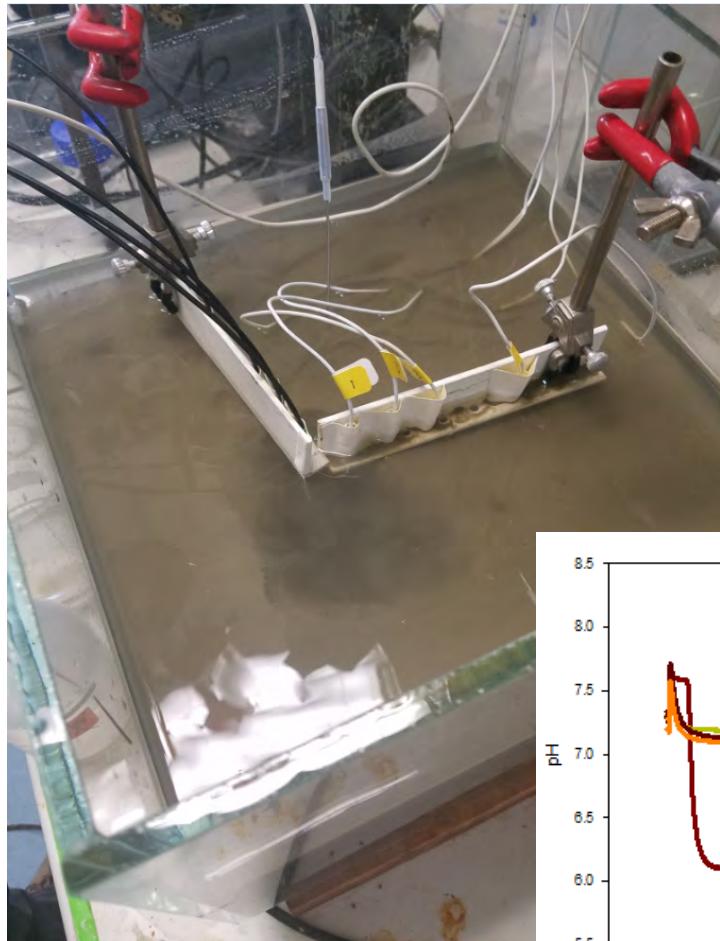
pH decrease local  
( $<1$  m from vent)

Variable &  
not proportional

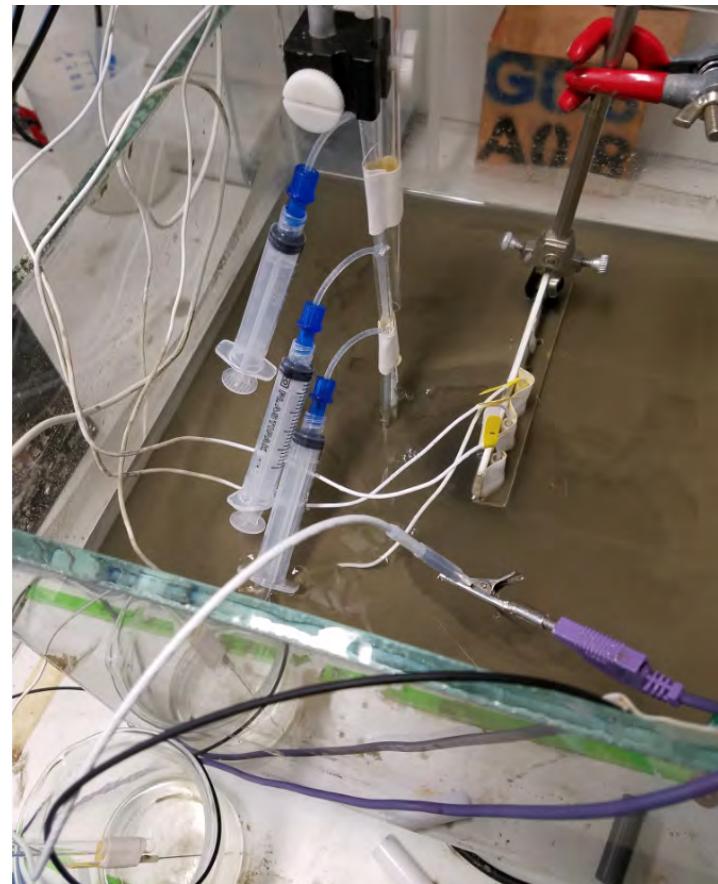
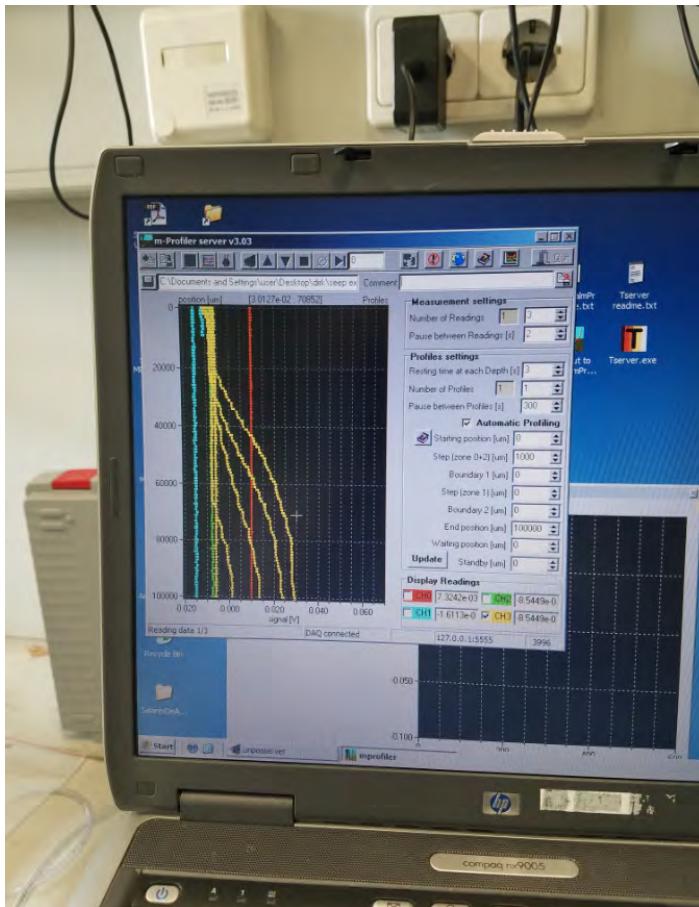
Local effect in  
bottom water



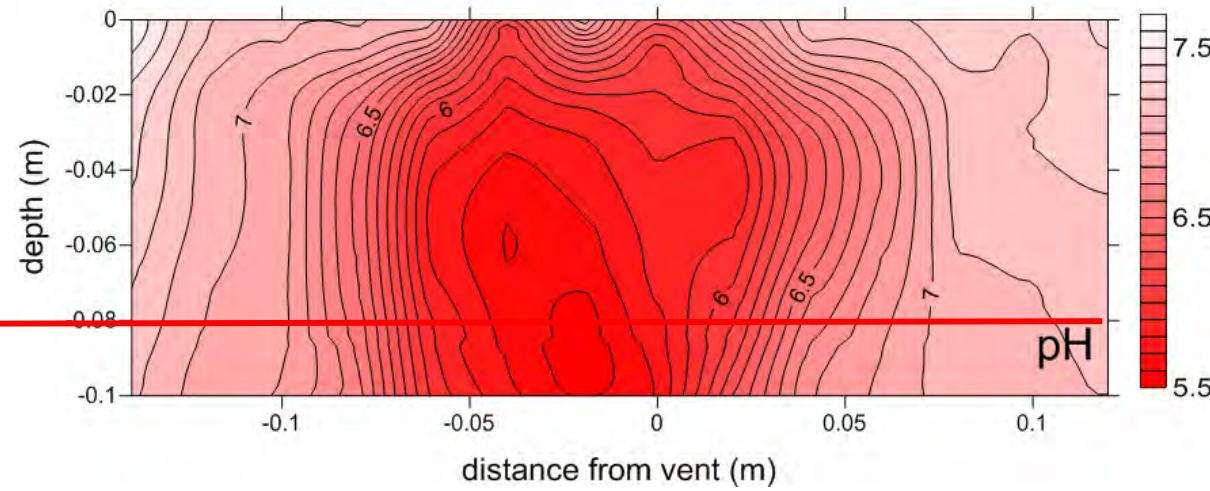
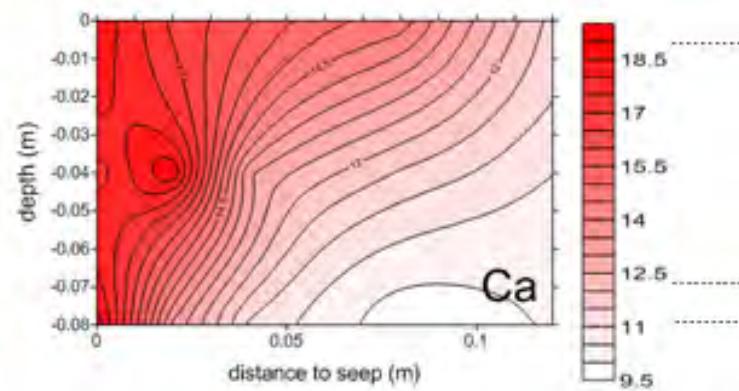
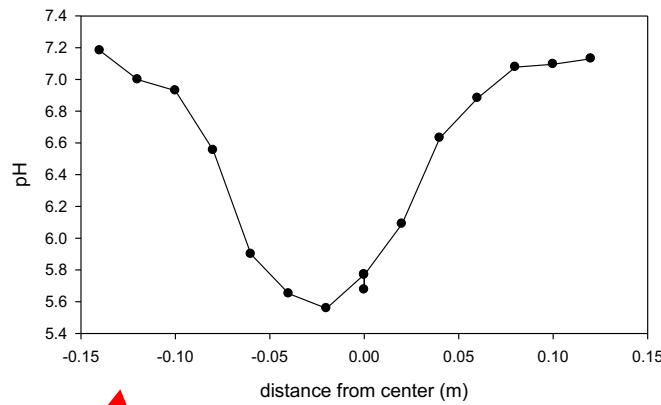
# Lab experiment



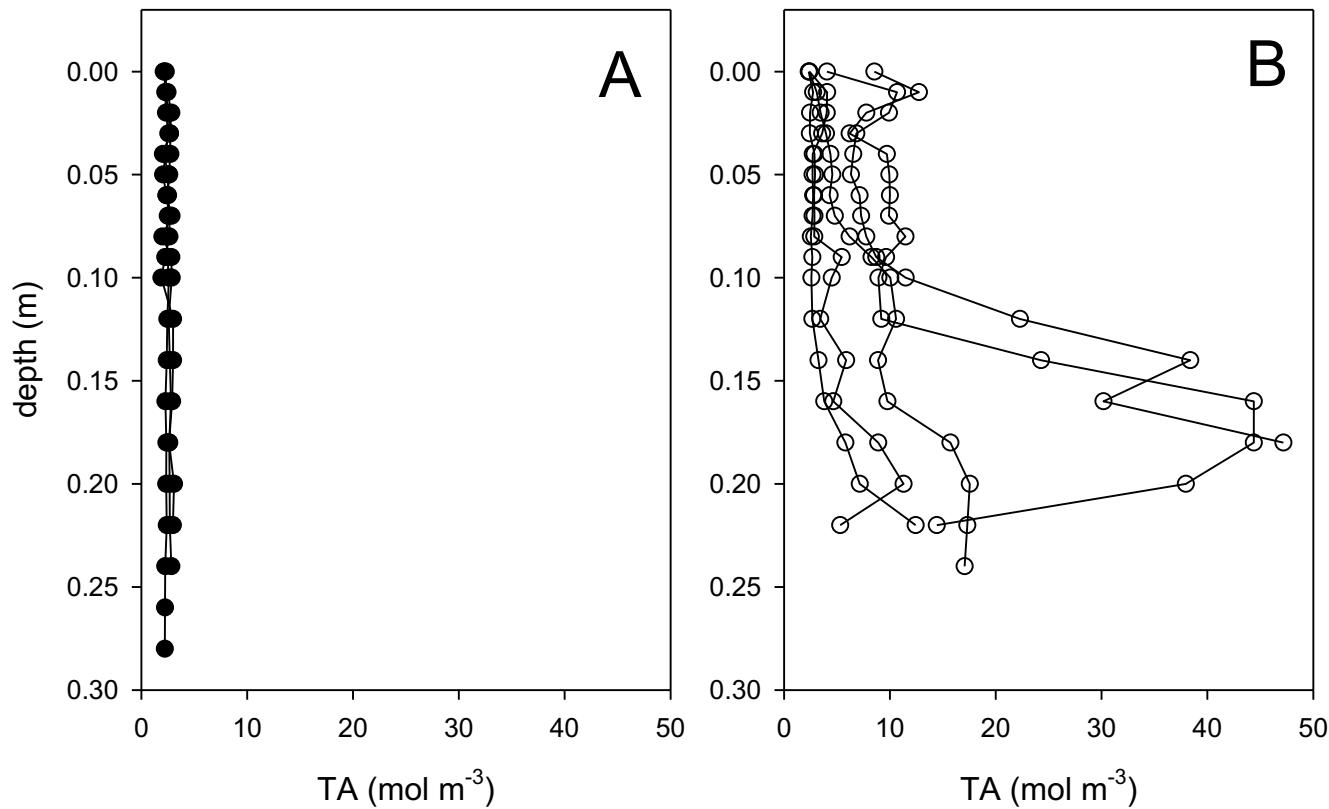
# At day 4



# Portrait of a vent



# TA values *in situ*

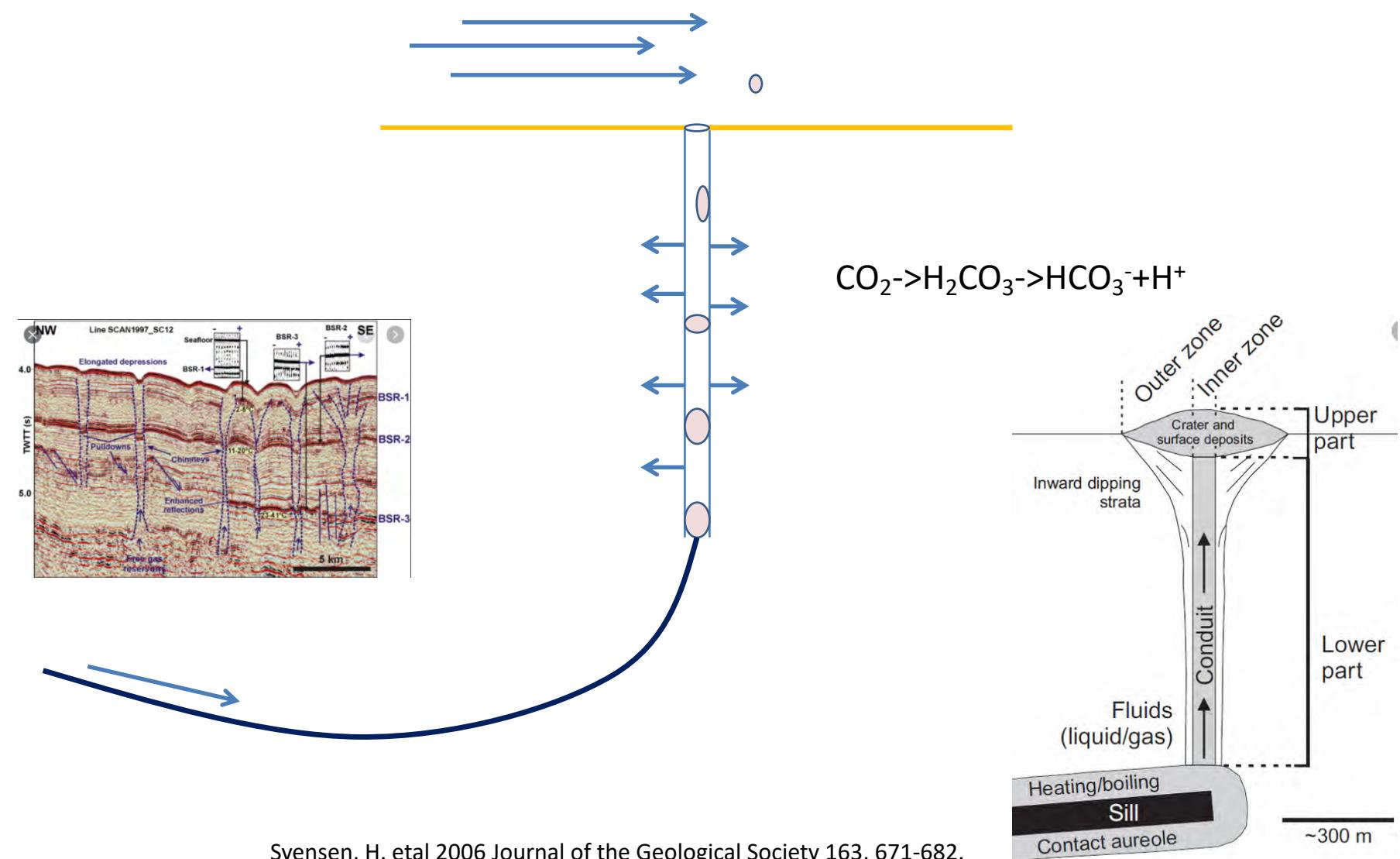


**Calcite dissolution (+silicate???)**

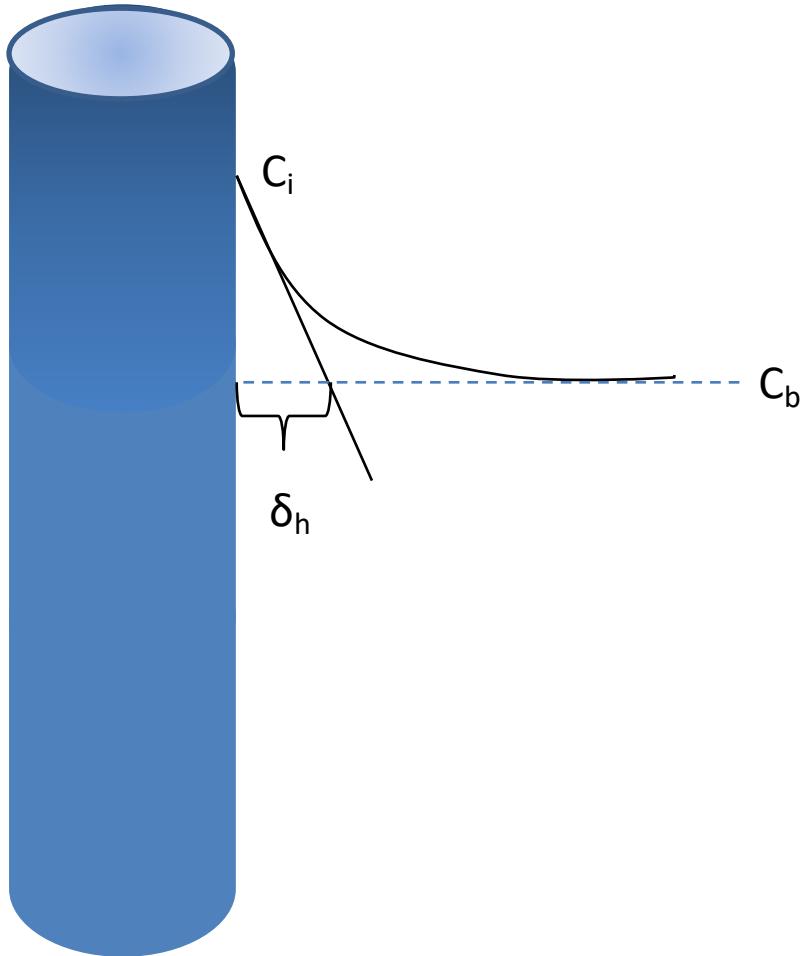
Kate Peel, Anita Flohr

Top 10 cm ventilated by escaping bubbles  
(Michael Stöhr: bubble out, water in)

# Vent=cylinder in infinite medium



# Cylindrical source: effects are local



$$J = k(C_b - C_i)$$

$k = D/\delta_h$   $k$ =mass transfer coefficient ( $m s^{-1}$ ),  
 $\delta_h$ =effective boundary layer

$$Sh = 0.3 + \frac{0.62 Re^{\frac{1}{2}} Sc^{\frac{1}{3}}}{\left(1 + \left(\frac{0.4}{Sc}\right)^{\frac{2}{3}}\right)^{\frac{1}{4}}} \left(1 + \left(\frac{Re}{282000}\right)^{\frac{5}{8}}\right)^{\frac{4}{5}}$$

$$= \frac{kd}{D}$$

In absence of flow  $Re=0$  thus:  $Sh = 0.3$

$$\text{As } Sh = \frac{kd}{D} \text{ and } k = \frac{D}{\delta_h} \quad \boxed{\delta_h = \frac{d}{Sh} = 3d}$$

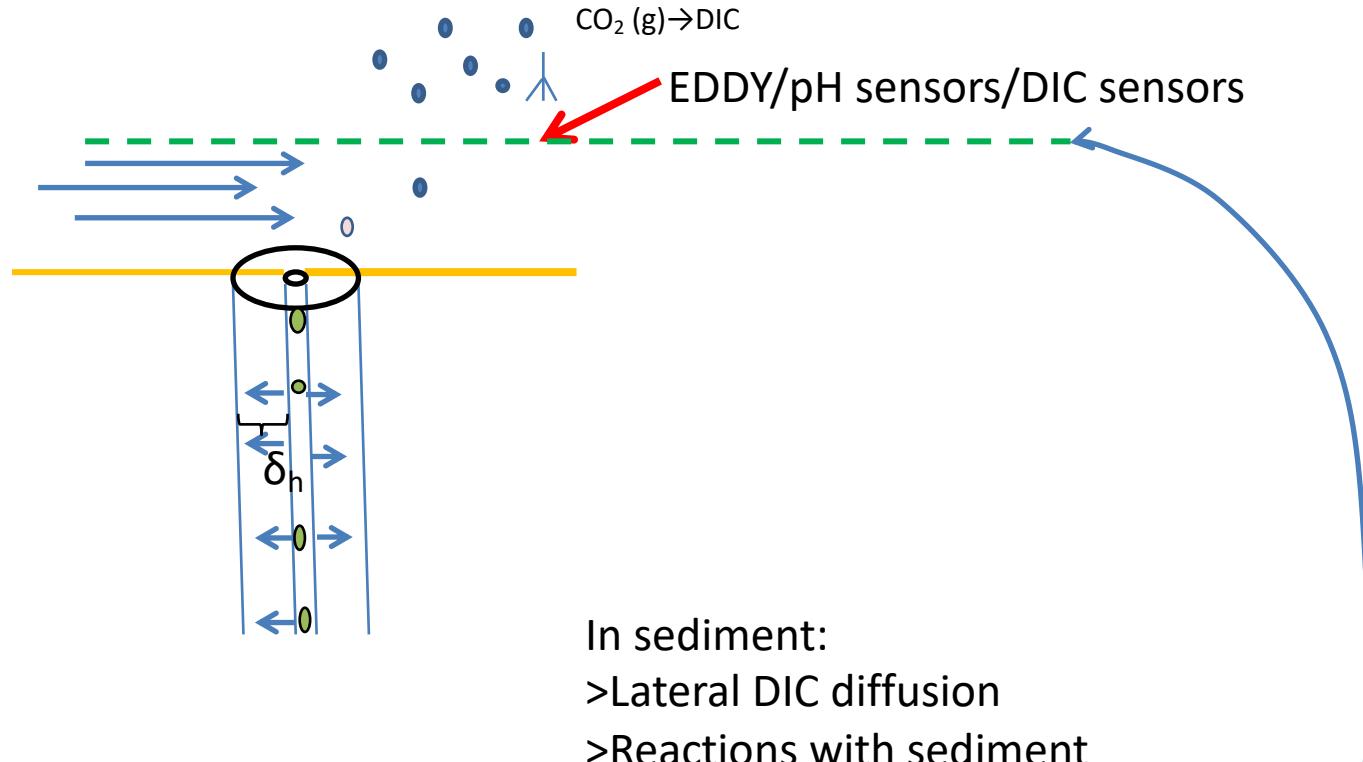
**In steady state and in absence of reactions:**

**Effective boundary layer is finite and about 3x the diameter**

$\delta_h$  independant from  $D$ , both for mass and heat

bubble~2 cm,  $\delta_h$ ~6 cm  
 diameter plume ~14 cm

# Mass loss (DIC)



In sediment:  
>Lateral DIC diffusion  
>Reactions with sediment

Above sediment:  
>Plume above measuring horizon  
> $\text{CO}_2(\text{l})$  ( $\text{CO}_2 \rightarrow \text{H}_2\text{CO}_3 \rightarrow \text{HCO}_3^- + \text{H}^+$  slow)

D? k?

mass transfer mechanism?

Pulsating channel....

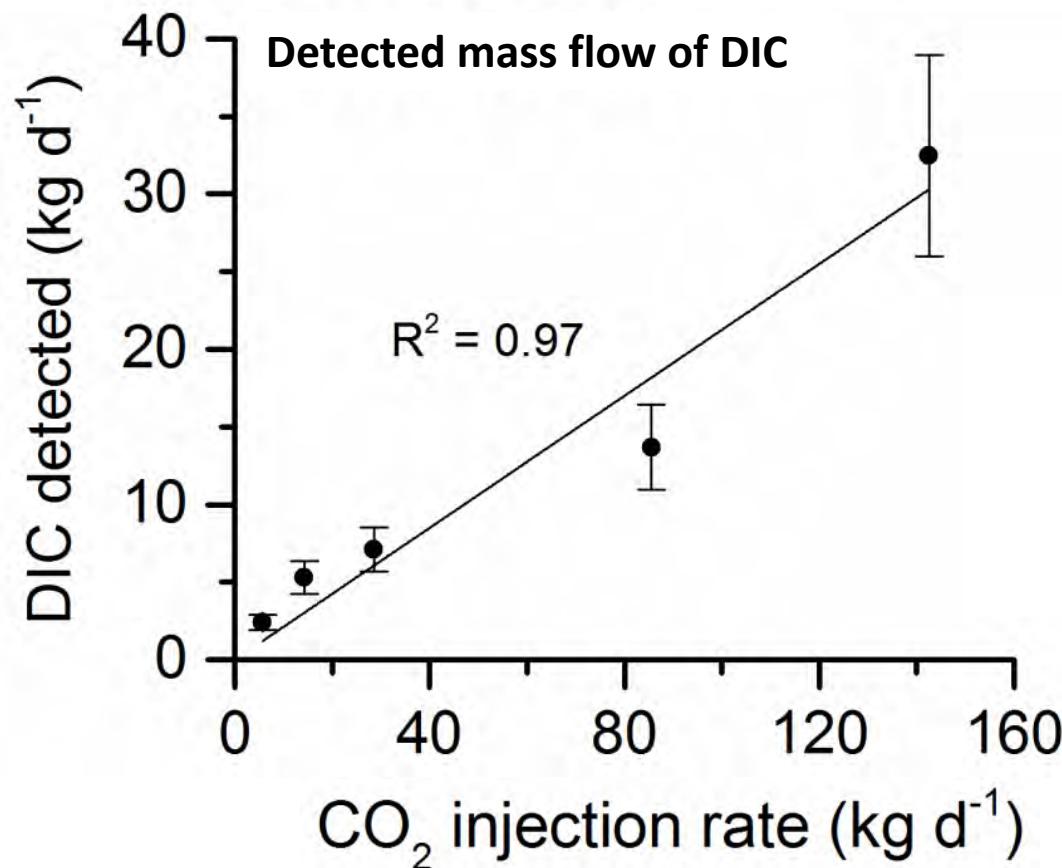
Carbonate equilibrium?

Kinetics of calcite dissolution?

Kinetics of silicate weathering?

# 75%?

- Poster Dirk Koopmans:
- 



# Heat generation

1.  $\text{CO}_2 (\text{g}) \rightarrow \text{CO}_2 (\text{aq})$   $\Delta H -20000 \text{ J mol}^{-1}$
2.  $\text{CaCO}_3 + \text{CO}_2 (\text{aq}) + 2\text{H}_2\text{O} \rightarrow \text{Ca}^{2+} + 2\text{HCO}_3^- + \text{H}_2\text{O}$   $\Delta H -10900 \text{ J mol}^{-1}$
3.  $\text{MeSiO}_3 + 2\text{CO}_2 (\text{aq}) + 4\text{H}_2\text{O} \rightarrow \text{MeSiO}_3 + 2\text{HCO}_3^- + 2\text{H}_3\text{O}^+ \rightarrow \text{Me}^{2+} + \text{H}_4\text{SiO}_4 + 2\text{HCO}_3^- + \text{H}_2\text{O}$   $\Delta H -73000 \text{ J mol}^{-1}$

Heat flux:  $3 \text{ J m}^{-2} \text{ s}^{-1}$

(from thermal conductivity and T gradient, underestimation!)

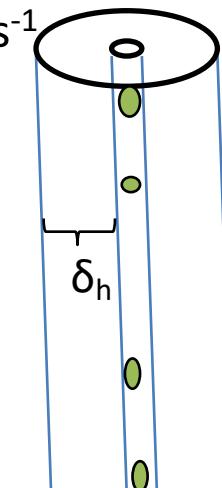
Assume cylinder of 3m long, 2 cm bubble channel plus 6 cm  $\delta_h$ :  $0.13 \text{ J s}^{-1}$

This heat is generated by:

$\text{CO}_2$  reactions 1+2:  $0.00013 \text{ mol s}^{-1}$

$\text{CO}_2$  reactions 1+3:  $0.00004 \text{ mol s}^{-1}$

2 L  $\text{CO}_2$  min:  $0.0015 \text{ mol s}^{-1}$



# Conclusions

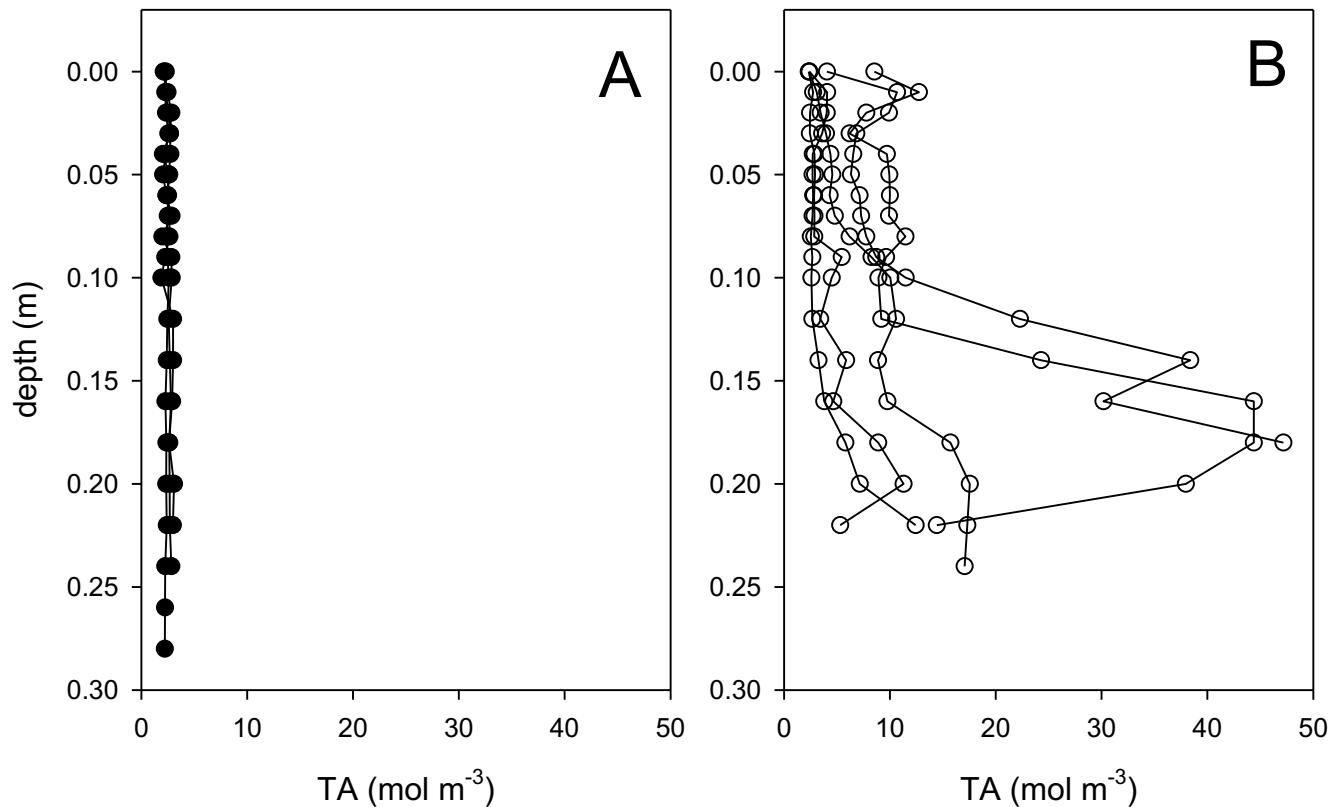
- Vent channels are narrow and have only local effects
  - MBL 3x diameter, if  $d=2$  cm MBL 6 cm total diameter influenced sediment  $2+2\times6=14$  cm
- $\text{CO}_2$  input generates DIC and heat
- Heat and DIC partially dissipated laterally in sediment
- **Heat** needs inclusion in vent **modeling** (kinetics and viscosity)



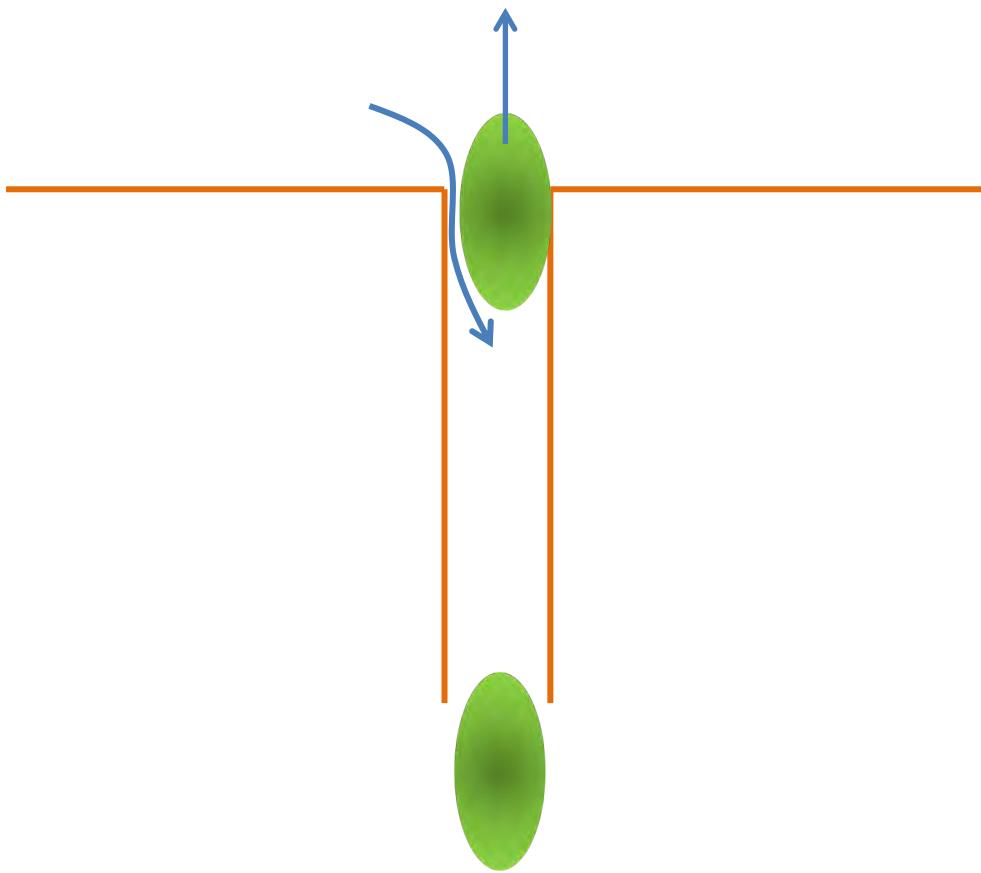




# TA values *in situ*

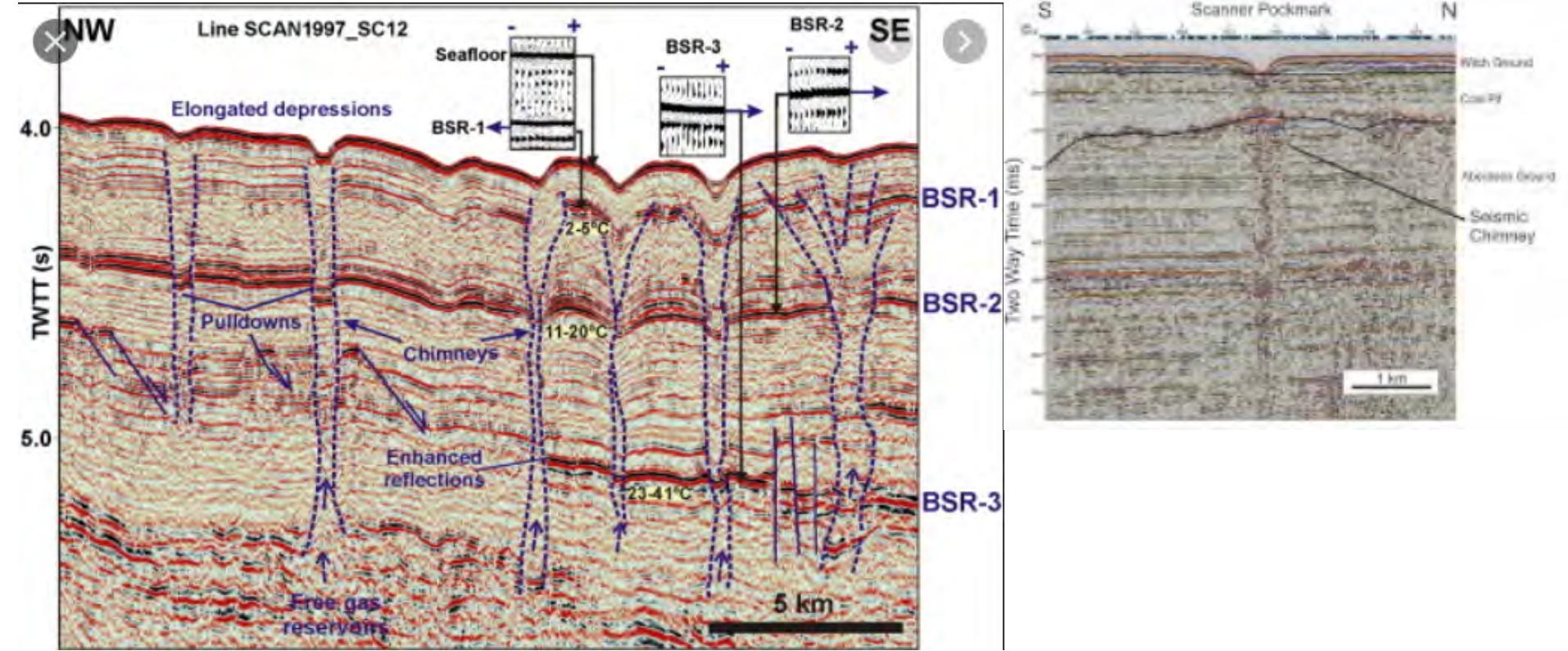


Very variable, top 10 cm ventilated by escaping bubbles  
(Michael Stöhr: bubble out, water in)

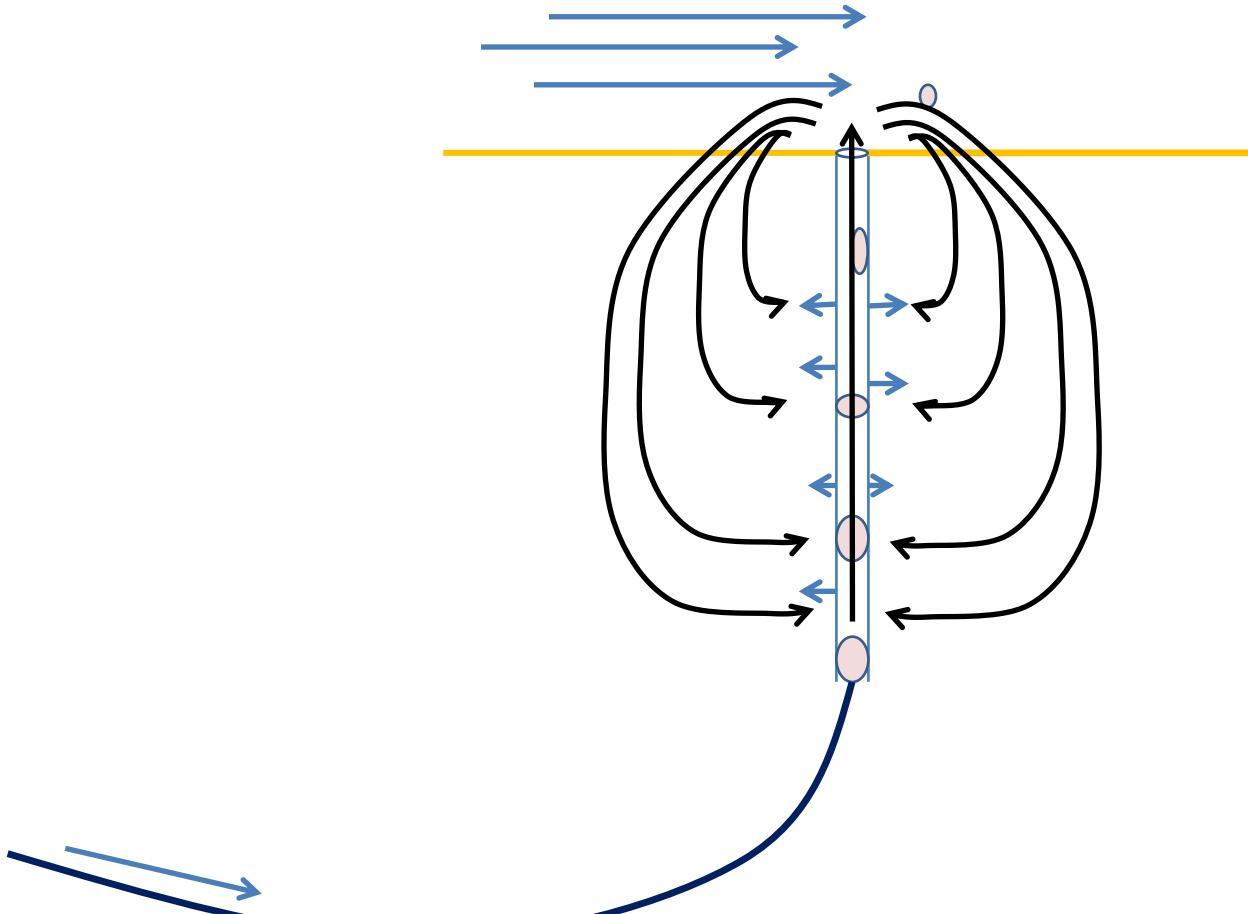


Stöhr et al, 2004

	mass		heat			
alk	50					
DIC in	10000					
J=	0.000166667 mol/m <sup>2</sup> s		3 J/m <sup>2</sup> s			
A=	0.4396		0.4396	L/min		
I=	3		3		2	0.001488 mol/s
Area cyl	1.3188		1.3188			
total flow	0.0002198 mol/s		3.9564 J/s			
		Calcite	0.000128039 mol/s			
		Silicate	4.25419E-05 mol/s			

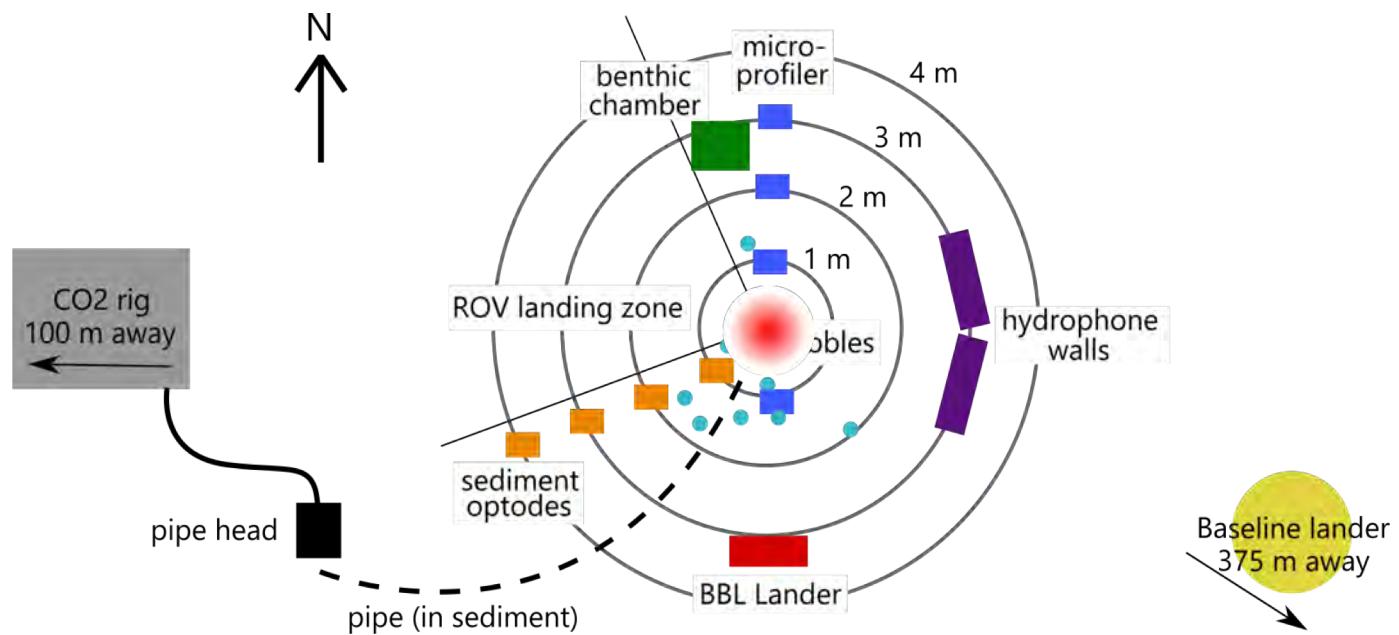


# Advection cell in vent



$Sh \uparrow$  thus  $\delta_h \downarrow$   
-> more local

Haeckel M, Boudreau BP, Wallmann K. 2007. Bubble-induced porewater mixing: A 3-D model for deep porewater irrigation. *Geochimica Et Cosmochimica Acta* **71**:5135-5154



# Ecology and geochemistry

Sulfate reduction lower  
(=35% total mineralization)

No effect O<sub>2</sub> uptake

pH decrease only local

No animals deterred

